

Novel Mono- and Multinuclear Porphyrazine Macrocycles. Physicochemical Properties and Perspectives as Multimodal Anticancer Agents

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Porphyrazines, apart from the extensively investigated phthalocyanines (tetrabenzoporphyrazines, Fig. 1A), have received increasing attention over the past decade. A lot of structural modifications are possible, one of which involves the annulated benzene rings of the phthalocyanine being replaced by electron-withdrawing heterocyclic rings containing S or N; this results in macrocycles having a general electron-deficient character (representative examples in Fig. 1B,C) [1]. These novel families of porphyrazines, while approaching in many respects the phthalocyanine framework, have distinct physicochemical behavior with implied differences in solid-state and/or solution, UV-visible and electrochemical properties. In addition, the presence of appropriate N-donor fragments in the periphery of the macrocycle allows the synthesis of multicationic water-soluble species and/or multimetallic macrocyclic systems with very interesting structural and electronic features in terms of applications [2]. In the recent decade, several members of the mentioned series of compounds, mainly carrying centrally Mg^{II} , Zn^{II} or Pd^{II} , have been especially studied as promising species for application in Photodynamic Therapy (PDT), a target widely used anticancer treatment, and in the multimodal anticancer therapy, an attractive field emerging in biomedical science [3].

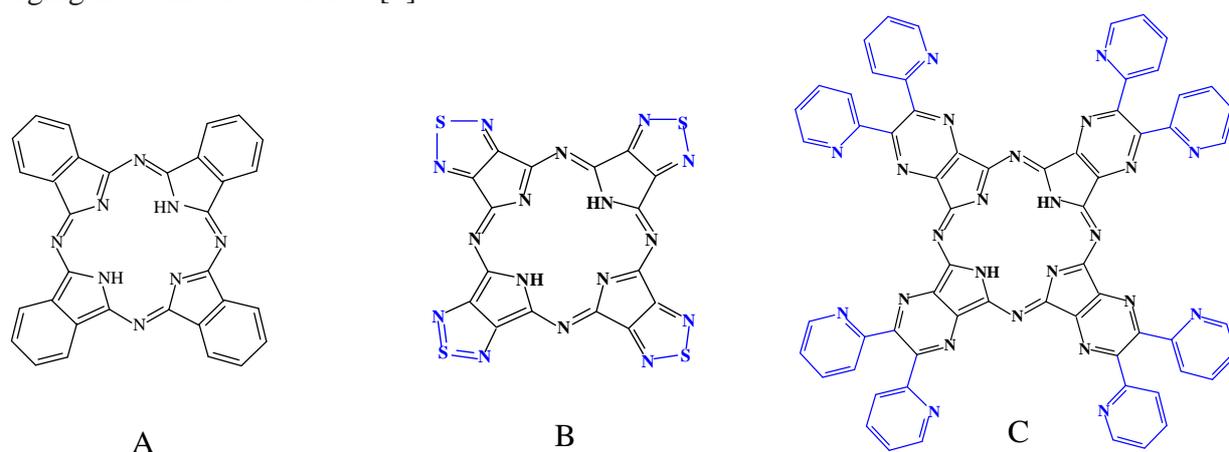


Figure 1. A: phthalocyanine; B: tetrakis(thiadiazole)porphyrazine;
C: tetrakis(dipyridinopyrazino)porphyrazine.

References

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